

Possibility of transition of hydrogen into the metallic state

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We investigated the resistivity of hydrogen at $T=4.2^\circ\text{K}$ in the pressure region $P \sim 1$ Mbar. A jump of the resistivity, amounting to six orders of magnitude, was observed. Observation of the "unfreezing" of the conducting state leads to the conclusion that the jump is connected with a phase transition.

The question of the transition of hydrogen into the metallic state has attracted much attention of late.^[1-7] According to estimates,^[1,3-7] this transition should occur at pressures of several megabars.

Observation of dielectric-metal transitions were previously reported^[8-11] in diamond,^[8,9] silica (SiO_2),^[10] Al_2O_3 , and other substances at pressures $P \sim 1$ Mbar.^[11]

In the present study, using a procedure similar to that of^[8-11], we have investigated the electric properties of hydrogen. To this end the hydrogen was precipitated on anvils cooled to 4.2°K and made of Carbonado-type diamond.^[12] The thickness of the hydrogen layer is regulated by the amount of condensed gas. In the case of relatively large thicknesses of the solid-hydrogen layer, the resistance (R) does not depend on the force (F) applied to the anvils and remains constantly high.

The constancy of the resistance R at $F \approx \infty$ points to the important fact that no appreciable outflow of solid hydrogen takes place from the gap between the anvils at the given temperature.

When thinner hydrogen layers are deposited, a jump of the resistance by at least six orders of magnitude is observed (Fig. 1).

To verify that the jump of the resistance is connected with a phase transition, we used the method of "unfreezing" the metastable phase. This method was used earlier in^[9-11].

Figure 2 shows a plot of the resistance in the case of repeated loading of the anvils. In this case the force was not dropped to a zero value, but to a value $F = F_b$, at which a conducting phase still existed, but symptoms

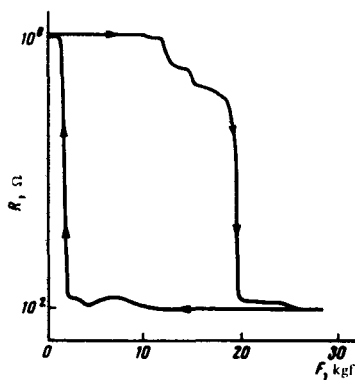


FIG. 1. Dependence of the resistance of hydrogen at $T=4.2^\circ\text{K}$ on the force applied to the anvils.

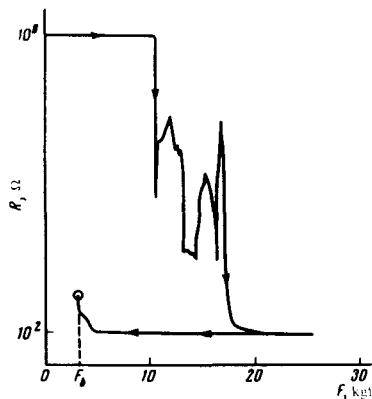


FIG. 2. Dependence of the resistance of hydrogen at $T=4.2^\circ\text{K}$ on the force following repeated loadings. The force was dropped to the value $F = F_b$.

of a transition to the dielectric state have already appeared. If the temperature is raised at a fixed force $F = F_b$, then the rate of the phase transition should increase, and the metastable metallic phase should "unfreeze."

Figure 3 shows the temperature dependence of the resistance at $F = F_b$.

Observation of the "unfreezing" effect leads to the conclusion that the jump of the resistance in hydrogen under pressure, under the conditions of our experiment, is due to a phase transition into the metallic state.

The pressure of the transition can presently be esti-

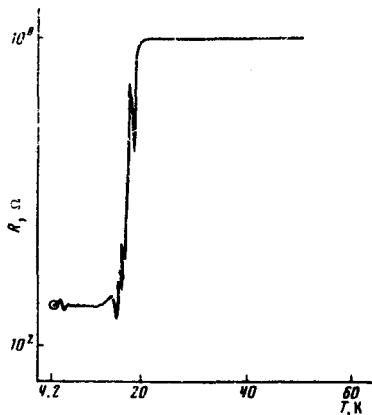


FIG. 3. Dependence of the resistance of hydrogen on the temperature at a fixed force F_b .

mated only approximately at $P \sim 1$ Mbar, which coincides in order of magnitude with the estimates given in^[1,3-7].

We note in conclusion that the return of the hydrogen to the dielectric state after the removal of the pressure (Fig. 1) cannot be regarded as a demonstration that metallic hydrogen cannot exist at normal pressure. As is well known, it is necessary to employ special measures to conserve the "quenching" of the metastable modifications of substances.

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