

INVESTIGATION OF STIMULATED EMISSION OF  $\text{CaF}_2\text{-Nd}^{3+}$  (TYPE II) CRYSTALS AT ROOM TEMPERATURE

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We have previously obtained stimulated emission at  $300^\circ\text{K}$  from  $\text{CaF}_2\text{-Nd}^{3+}$  (type I) at a wavelength  $\sim 1.047\mu$ .<sup>[1]</sup> In this communication we report preliminary results of an investigation of generation at  $\lambda \approx 1.088\mu$  and  $300^\circ\text{K}$  from the  $\text{Nd}^{3+}$  ion in  $\text{CaF}_2$  crystals (type II). Unlike type I, crystals of type II contain oxygen ions in the neodymium optical centers.

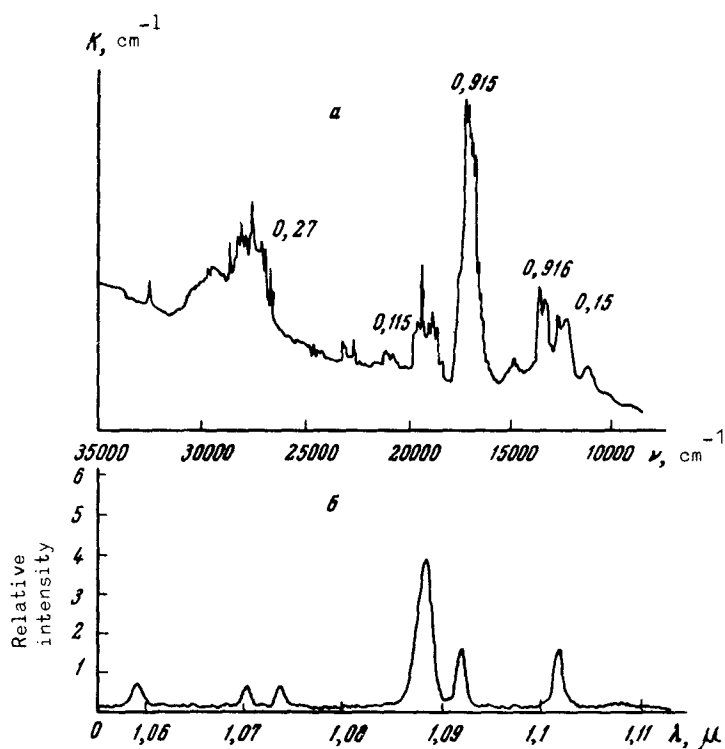


Figure 1

a - Absorption spectrum of  $\text{CaF}_2\text{-Nd}^{3+}$  (type II) at  $300^\circ\text{K}$

b - luminescence spectrum of the  ${}^4\text{F}_{3/2} \rightarrow {}^4\text{I}_{11/2}$  transition in the same crystal at  $300^\circ\text{K}$

We used  $\text{CaF}_2$  crystals with  $\text{Nd}^{3+}$  concentrations from 0.2 to 0.5%. Figure 1a shows the absorption spectrum of type II  $\text{CaF}_2\text{-Nd}^{3+}$ , obtained at  $300^\circ\text{K}$  with an SP-700 instrument. The  $\text{Nd}^{3+}$  concentration is approximately 0.5%. Some of the characteristic lines are marked with their absorption coefficients. The system was excited in an absorption band lying 20 000  $\text{cm}^{-1}$  lower, using an IFP-800 xenon lamp. The infrared luminescence of the

$\text{Nd}^{3+}$  ion begins at the  ${}^4\text{F}_{3/2}$  level and ends at different levels of the  ${}^4\text{I}$  ground state multiplet. The strongest luminescence corresponds to the  ${}^4\text{F}_{3/2} \rightarrow {}^4\text{I}_{11/2}$  transition, just the one in which generation was attained. Figure 1b shows the luminescence spectrum of the  ${}^4\text{F}_{3/2} \rightarrow {}^4\text{I}_{11/2}$  transition of the  $\text{Nd}^{3+}$  ion in type II  $\text{CaF}_2$ , obtained at  $300^\circ\text{K}$  with a DFS-12 instrument. The  $\text{Nd}^{3+}$  concentration is approximately 0.2%.

We used in the experiments an elliptical illuminating system with optical efficiency  $\sim 0.5$ .<sup>[2]</sup> The working crystals were cylindrical rods with polished ends parallel to within  $\sim 15''$ . The crystals were  $\sim 6.0$  mm in diameter and 75 mm long. The optical cavity was made up of confocal dielectric mirrors with  $\sim 2\%$  transmission at  $1.06 \mu$ . The mirrors were 40 mm in diameter and of 500 mm curvature radius.

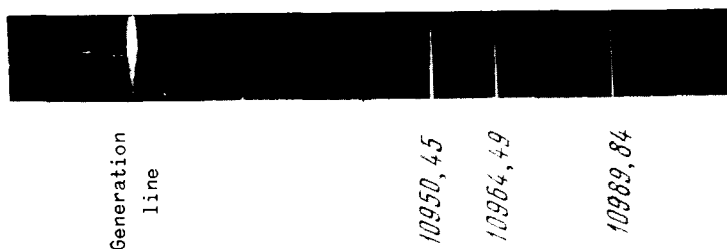


Figure 2  
Stimulated emission spectrum  
of  $\text{CaF}_2\text{-Nd}^{3+}$  at  $300^\circ\text{K}$

Figure 2 shows the emission spectrum of  $\text{CaF}_2\text{-Nd}^{3+}$  (type II), obtained with a DFS-8 instrument with 600 lines/mm grating. The dispersion of the instrument in the  $1.1 \mu$  region is  $5.99 \text{ \AA}/\text{mm}$ . The reference was the third-order spectrum of the PRK-2 lamp. The generation was registered with I-1070 film. At  $300^\circ\text{K}$  the width of the emission line is  $\sim 0.9 \text{ \AA}$  ( $0.76 \text{ cm}^{-1}$ ). The lifetime of the excited state  ${}^4\text{F}_{3/2}$  of the  $\text{Nd}^{3+}$  ion in  $\text{CaF}_2$  (type II) was measured at  $300^\circ\text{K}$  with a tau-meter developed by us, and amounts to  $\sim 1.25$  msec for the generating samples.

To observe the time dependence of the emission, we used a photomultiplier with oxygen-cesium photocathode and with a 1-mm filter at its input. The photomultiplier signal was fed to an oscilloscope. The recording system had a time constant  $\sim 10^{-6}$ .

Figure 3 shows oscillograms of the emission pulses from  $\text{CaF}_2\text{-Nd}^{3+}$  (type II) at threshold excitation energy and with the threshold exceeded by a factor of 3.

No.	Substance	$\lambda, \mu$	Threshold elec. energy, J	Excited state lifetime, msec.	Ref.
1	$\text{Cr}^{3+} - \text{Al}_2\text{O}_3$	0.6934		3	[3]
2	$\text{Nd}^{3+} - \text{SrF}_2$	1.0370	480		[4]
3	$\text{Nd}^{3+} - \text{CaF}_2$ type I	1.047	130	~1.0	[1]
4	$\text{Nd}^{3+} - \text{SrMoO}_4$	1.0576	45		[5]
		1.0648	125		
5	$\text{Nd}^{3+} - \text{CaWO}_4$	1.0582	2	~0.2	[6]
		1.0652	3		
6	$\text{Nd}^{3+} - \text{PbMoO}_4$	1.0586	60		[4]
7	$\text{Nd}^{3+} - \text{SrWO}_4$	1.063	180		[4]
8	$\text{Nd}^{3+} - \text{LaF}_3$	1.0633	150		[4]
9	$\text{Nd}^{3+} - \text{CaMoO}_4$	1.0673	360		[4]
IO	$\text{Nd}^{3+} - \text{glass}$	~1.06	~50		[7]
II	$\text{Nd}^{3+} - \text{Y}_3\text{Al}_5\text{O}_{12}$	1.0648	2	0.2	[8]
I2	$\text{Nd}^{3+} - \text{Y}_3\text{Ga}_5\text{O}_{12}$	1.0633	250	0.2	[8]
I3	$\text{Nd}^{3+} - \text{Gd}_3\text{Ga}_5\text{O}_{12}$	1.0633	350	0.2	[8]
I4	$\text{Nd}^{3+} - \text{CaF}_2$ type II	1.0885	170	1.25	Present work
I5	$\text{Nd}^{3+} - \text{Y}_2\text{O}_3$	1.073	260	~0.26	[9]
		1.078	350	~0.26	
I6	$\text{Nd}^{3+} - \text{Ga}_2\text{O}_3$	1.074I	~36	~0.12	[10]
		1.0789	~9	~0.12	

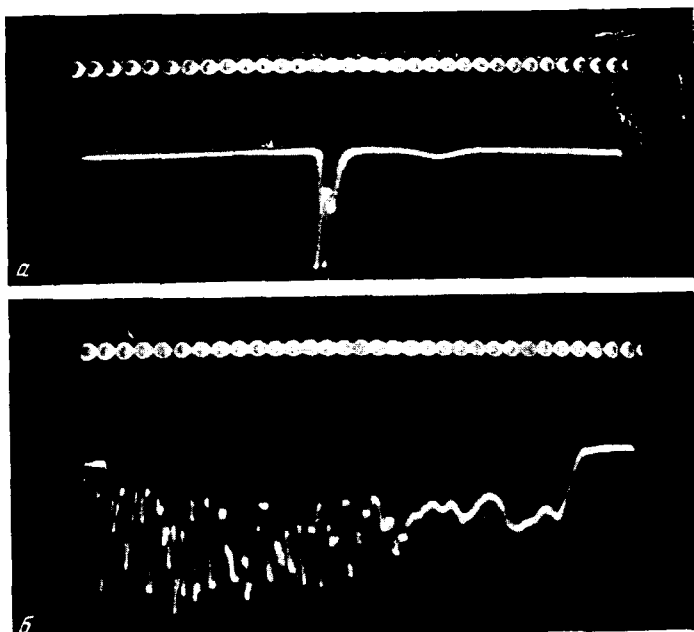


Figure 3  
 Time dependence of stimulated emission: a - at threshold excitation energy, b - at triple the excitation energy. The time markers indicate 20  $\mu$ sec intervals.

The table lists some characteristics of lasers operating at 300°K. It shows that the laser with  $\text{CaF}_2\text{-Nd}^{3+}$  (type II) has the longest wavelength of all known neodymium lasers.

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