GENERATION SPECTRUM OF A CHEMICAL LASER USING A MIXTURE OF H, AND Cl,

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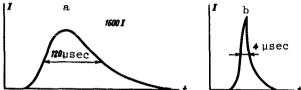
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The use of exothermal reactions for the development of chemical lasers is quite promising at present.

An experimental paper by Kasper and Pimental [1] describes generation in a mixture of  $H_2$  and  $Cl_2$  gases at the HCl vibrational transitions. The generation spectrum was related to transitions between the levels v = 1 and v = 0, and was attributed to the presence of partial inversion at the vibrational-rotational levels.

In early 1968 we also investigated lasing by this mixture. The experimental setup was similar to that used in [1]. One of the typical time patterns of the lasing and of the pump pulses is shown in the figure.

Unlike the results of [1], where the generated radiation was connected with the vibrational transitions  $v=1 \rightarrow v=0$ , our experiment has shown that the emission can be identified with the transition  $2 \rightarrow 1$ . This is important for the estimate of the efficiency of the elementary act  $H + Cl_2 = HCl^+$ +  $Cl_1$ , in other words, of that fraction of



a) Time dependence of the pump pulse, b) time dependence of the 2652 cm<sup>-1</sup> generation line (15 Torr Cl<sub>2</sub>, 40 Torr H<sub>2</sub>, T = 300°K).

the energy which is released during the course of the reaction and is stored in the oscillatory reservoir and can be used in principle to obtain generation.

Polanyi [2] investigated the energy distribution of the HCl molecules obtained as a result of the elementary act, and showed that total inversion occurs in the vibrational transitions  $v = 2 \rightarrow v = 1$  and  $v = 3 \rightarrow v = 2$ ; he could not understand why this inversion could not be realized in [1]. To explain the results of [1] it was necessary to assume too rapid a process of establishment of the Boltzmann distribution over the vibrational levels. Observation of generation at the transition  $v = 2 \rightarrow v = 1$  eliminates this contradiction and is of importance for the understanding of the kinetics of inversion during the course of the chemical reaction between the hydrogen and the chlorine.

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- [2] J. C. Polanyi, J. Chem. Phys. 34, 347 (1961).

## SURFACE EFFECTS IN A STRONG MAGNETIC FIELD

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There are two well-known types of skin effect: normal, when the skin-layer depth  $\delta$  is large compared with the mean free path  $\ell$  of the charges, and anomalous when  $\delta << \ell$ . In either