

# Complicated exciton-impurity complexes in tin oxide

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(Submitted April 12, 1976)

Pis'ma Zh. Eksp. Teor. Fiz. 23, No. 10, 574-577 (20 May 1976)

When  $\text{SnO}_2$  is doped with indium or antimony, complex centers are produced, to which excitons start to be bound. The ground state of the exciton-impurity complex yields an absorption line whose energy does not depend on the type of introduced impurity, whereas the excited states have different spectra.

PACS numbers: 78.50. — w

1. In some not-specially-doped  $\text{SnO}_2$  crystals, lines of exciton-impurity complexes (EIC) with low exciton-impurity binding energy are observed on the long-wave side of the absorption line of the free exciton. These lines are located  $\sim 15 \text{ cm}^{-1}$  away from the 1S line and probably correspond to the shallow donors typical of  $\text{SnO}_2$  (the Rydberg constant of the exciton series is equal to  $270 \text{ cm}^{-1}$ ). The spectra of the doped crystals contain EIC absorption lines with much larger binding energy and with specific distinguishing properties.

2. Whereas the 1S state of the free excitons is quadrupole,<sup>[1]</sup> all the EIC lines adjacent to it, even the shallowest, are dipole. The relative intensity of the EIC lines in  $\text{SnO}_2$  are determined both by the giant oscillator strength<sup>[2]</sup> and by the allowed type of the optical transition. Consequently, the EIC lines at the doping level  $10^{17} \text{ cm}^{-3}$  are more intense than the 1S-state lines of the free exciton. The dipole character of the EIC lines manifests itself in their identical intensity for any position of the wave vector  $\mathbf{k}$  of the light in a plane perpendicular to the fourfold axis  $C_4$ , whereas the quadrupole 1S line oscillates in the case of rotation in this plane (Fig. 1).

3. After "pure"  $\text{SnO}_2$  crystals are doped by the thermal diffusion method, an EIC line 0' appears in their spectrum, shifted  $70 \text{ cm}^{-1}$  away from 1S, as well as a group of narrow absorption lines on the long-wave side of  $2P_{41}$ —the lines

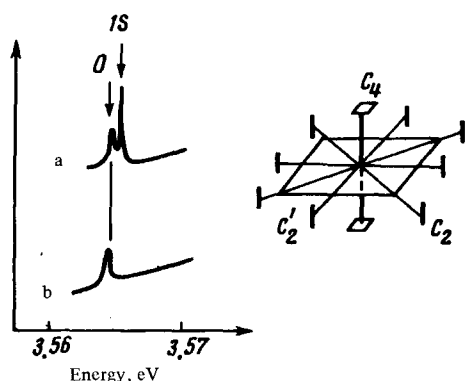


FIG. 1. Free exciton 1S and shallow bound exciton (0) in the absorption spectrum of  $\text{SnO}_2$  when observed along the twofold axes  $C_2(a)$  and  $C_2'(b)$ .  $T = 4^\circ\text{K}$ . Right—scheme of the symmetry axes of  $\text{SnO}_2$ .

1–6 of the free exciton. These lines correspond to excited EIC states.<sup>1)</sup> It would be of interest to analyze these states for the construction of a molecular model of the EIC, since the ground state has only one level and does not afford such an opportunity. However, the lines 1–6 may correspond, first, to  $nP_{\pm 1}$  excitons, where  $n$  need not necessarily be equal to 2, and second, they can correspond to states  $nP_0$ ,  $nS$ ,  $nD$ , etc., since forbidden transitions to the free exciton levels correspond to dipole transitions to EIC levels. If the binding of the exciton to the impurity can be regarded as a perturbation, then for a more detailed analysis of the EIC it is necessary to know the dependence of the energy of EIC production on the exciton quantum numbers  $n$  and  $l$ , a dependence which apparently has not been calculated. For a correct analysis it is necessary also to estimate the ratio of the exchange forces and the Coulomb forces leading to the formation of the EIC in the case of neutral and charged defects, and the Coulomb forces in the exciton. We note that the distance of lines 1 and 2 (Fig. 2) from the longest-wavelength  $2P_{\pm 1}$  line<sup>2)</sup> of the free exciton is larger than the distance between the  $0'$  and  $1S$  lines, and these energy distances exceed the binding energy of the  $2P_{\pm 1}$  state of the free exciton.

4. Doping of identical "pure"  $\text{SnO}_2$  crystals with indium or antimony yields identical lines of the ground state of EIC. The energies and relative intensities of lines 5 and 6 of the excited states are equal, and those of the remaining lines are not. The indium and antimony replacing the tin become respectively an acceptor and donor, and therefore the EIC spectra should differ strongly. The fact that in both cases the ground-state lines of the EIC are exactly identical indicates that the doping produces not a simple substitution but the formation of a more complex center, consisting of an atom of the introduced impurity and of a previously existing defect. Identical EIC are produced in crystals grown by different methods, and therefore the initial is apparently an intrinsic one. The reason why the EIC spectrum appears only after doping may lie both in the very nature of the complex center produced in this case, and in the consolidation of the intrinsic defect, which decreases the lattice distortion near it.

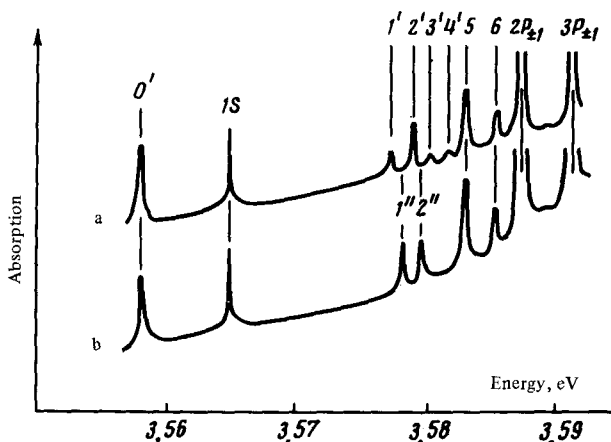


FIG. 2. Absorption spectra of free excitons  $1S$ ,  $2P_{\pm 1}$ ,  $3P_{\pm 1}$ , and EIC in the crystals  $\text{SnO}_2:\text{Sb}$  (a) and  $\text{SnO}_2:\text{In}$  (b).  $T = 4^\circ\text{K}$ .

Thus, a characteristic property of the exciton bound with a complex center is the fact that the binding energy of the ground state of the EIC is determined only by the initial defect, whereas the spectrum of the excited states of the EIC depends on the type of introduced impurity. It is this which offers evidence of the complexity of the center binding the exciton.

5. In a magnetic field  $H$ , numerous very narrow lines appear in the spectrum of tin dioxide, some of which do not pertain to the states of the free exciton. These lines can be interpreted as an EIC spectrum corresponding to free excitons with large values of  $n$ , which become stabilized in the magnetic field, that is, the magnetic-absorption spectrum of semiconducting crystals with EIC centers. In addition, besides the series of the diamagnetic excitons adjacent to different band levels, also EIC lines. With increasing  $H$ , the energy difference between the EIC lines and the free-exciton lines should decrease, since the external perturbation predominates in a strong field, and the influence of the binding of the exciton to the impurity becomes negligible.

Lines 1—6 cannot be phonon replicas of the ground state of an EIC, since the phonons interacting with it have a much higher energy.<sup>[3]</sup>

The 2S line in  $\text{SnO}_2$  has a higher energy than  $2P_{41}$ .

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